

## POLYATOMIC IONS CHART

Polyatomic Ion	Chemical Symbol	Oxidation Number
<i>ammonium</i>	$NH_4^+$	1+
acetate	$C_2H_3O_2^-$	1-
bicarbonate	$HCO_3^-$	1-
bisulfate	$HSO_4^-$	1-
bisulfite	$HSO_3^-$	1-
bromate	$BrO_3^-$	1-
Bromite	$BrO_2^-$	1-
carbonate	$CO_3^{2-}$	2-
chlorate	$ClO_3^-$	1-
chlorite	$ClO_2^-$	1-
chromate	$CrO_4^{2-}$	2-
cyanide	$CN^-$	1-
dichromate	$Cr_2O_7^{2-}$	2-
ferrocyanide	$Fe(CN)_6^{4-}$	4-
ferricyanide	$Fe(CN)_6^{3-}$	3-
hydroxide	$OH^-$	1-
hypobromite	$BrO^-$	1-

Polyatomic Ion	Chemical Symbol	Oxidation Number
hypochlorite	$ClO^-$	1-
hypoiodate	$IO^-$	1-
iodate	$IO_3^-$	1-
Iodite	$IO_2^-$	1-
nitrate	$NO_3^-$	1-
nitrite	$NO_2^-$	1-
oxalate	$C_2O_4^{2-}$	2-
perbromate	$HBrO_3^-$	1-
perchlorate	$ClO_4^-$	1-
permanganate	$MnO_4^-$	1-
peroxide	$O_2^{2-}$	2-
phosphate	$PO_4^{3-}$	3-
sulfate	$SO_4^{2-}$	2-
sulfite	$SO_3^{2-}$	2-
thiosulfite	$S_2O_3^{2-}$	2-
hydrogen phosphate	$HPO_4^{2-}$	2-
dihydrogen phosphate	$H_2PO_4^-$	1-